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- 2 - Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, $\text{HOOC}-\text{COOH}\cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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(b) The hydrolysis of 2-bromo-2-methylbutane by 0.10 mol dm^{-3} potassium hydroxide solution to form 2-methylbutan-2-ol is an example of an $\text{S}_{\text{N}}1$ reaction. (i) Explain how this $\text{S}_{\text{N}}1$ reaction occurs, using curly arrows to represent the movement of electron pairs, and showing the structure of the organic intermediate. [4]

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- 2 - Answer all questions. Write your answers in the boxes provided. 1. The rate of the acid-catalysed iodination of propanone can be followed by measuring how the

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(only 2 significant figures) because of precision of time/[Br 2] measurements; [2] Allow answers based on rate laws or orders of reaction. M2 can only be scored if M1 correct. (iii) [Br 2]: 1% and Time: 0.8%; Percentage Uncertainty: 1.8%; [2] Accept Percentage Uncertainty: 2%. Do not allow answers based on rate laws or orders of reaction. (b ...

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