

# Bohr Model Of Hydrogen Gizmo Answer Sheet

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The Bohr model of the atom was inspired by the spectrum produced by hydrogen gas. The link between light and atomic structure is illustrated by the Bohr Model of Hydrogen Gizmo. The Gizmo shows a laser pointed at a container of hydrogen gas. After passing through the gas, light from the laser goes through a prism and is detected on a screen.

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Bohr Model: Introduction. Launch Gizmo. Fire photons to determine the spectrum of a gas. Observe how an absorbed photon changes the orbit of an electron and how a photon is emitted from an excited electron.

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Bohr model of hydrogen Figure 31: Democritus The atomic theory of matter has a long history, in some ways all the way back to the ancient Greeks (Democritus - ca 400 BCE - suggested that all things are composed of indivisible \atoms") From what we can observe, atoms have certain properties and behaviors, which can be summarized as

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Gizmo Warm-up In 1913, Niels Bohr proposed that the unique spectral lines created by different elements were related to the way electrons were arranged around the nucleus. The Bohr Model:

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Introduction Gizmo™ explores this connection. The laser shown in the Gizmo can emit photons, or particles of light, at a variety of wavelengths.

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Chapter 3 Bohr model of hydrogen Figure 3.1: Democritus The atomic theory of matter has a long history, in some ways all the way back to the ancient Greeks (Democritus - ca. 400 BCE - suggested that all things are composed of indivisible "atoms").

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absorption spectrum of hydrogen. An emission spectrum is created when hydrogen gas emits light. The Bohr model of the atom was inspired by the spectrum produced by hydrogen gas. The link between light and atomic structure is illustrated by the Bohr Model of Hydrogen Gizmo.

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Although revolutionary at the time, the Bohr model is a relatively primitive model of the hydrogen atom compared to the valence shell atom. As an initial hypothesis, it was derived as a first-order approximation to describe the hydrogen atom.

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